

Strategic engineering for detecting antimicrobial compounds from *Taxus wallichiana* Zucc. (Himalayan yew)

Priyanka Adhikari¹, Vasudha Agnihotri^{2*} and Anita Pandey^{3*} 

¹ Central Ayurveda Research Institute, Bidhannagar- 700091, Kolkata, West Bengal, India

² G.B. Pant National Institute of Himalayan Environment-Himachal Pradesh Regional Centre, Mohal-Kullu, 175126, Himachal Pradesh, India

³ Department of Biotechnology, Graphic Era (Deemed to be University), Clement Town, 248002, Dehradun, Uttarakhand, India

* Corresponding authors, E-mail: vasudha@gbpihed.nic.in; anitapandey333@gmail.com

Abstract

Taxus wallichiana Zucc. (Himalayan yew) has been well-documented for containing therapeutically significant active ingredients. Its bark contains pharmaceutically important compounds i.e., taxol and its derivatives which are well known for their anticancer potential. However, *T. wallichiana* has received limited attention for its equally significant antimicrobial properties. Keeping this background in view, *T. wallichiana* was selected for the detailed investigation of antimicrobial activities, and isolation and characterization of secondary metabolites responsible for antimicrobial activity in different plant parts i.e., needle, bark, and stem extracts. In plate-based bioassays, plants exhibited antimicrobial action against the three main categories of microorganisms (fungi, bacteria, and actinobacteria). Based on the preliminary antimicrobial study, methanol and ethyl acetate extracts, were selected for further experiments. The bioautographic technique was used for identification, and the mobile phase was optimized with the help of a selectivity triangle. After continuous column and thin-layer chromatography, fractions were identified as having good antifungal, antibacterial, and antiactinobacterial activity. These fractions were selected for further characterization using techniques like GC-MS/LC-MS, and FTIR. These analyses support the identification of several fatty acids, including arachidic acid, behenic acid, palmitic acid, and stearic acid; vitamins (nicotinamide); alkaloids (cinchonine, timolol); amino benzamides (procainamide); carbocyclic sugars (myoinositol); and alkane hydrocarbons (hexadecane), which have antimicrobial activity in *T. wallichiana* needles. The information gathered from this study will help modern medicine make new drug discoveries that combine different active ingredients from medicinal plants to treat a wide range of ailments.

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Introduction

Microorganisms are developing multidrug resistance (MDR) for many antibiotic and medicinally important compounds due to improper usage. This is causing a major threat to society, the economy, and public health^[1]. Therefore, alternate sources of drugs are being searched which could be suitable for the microbe's MDR properties^[2]. There have been documented efforts in research to identify new, powerful natural or synthetic antimicrobial agents to combat MDR^[3,4]. Recently, it was shown that synthetic cationic and hydrophobic peptides with N-terminal labels, which are linked to the human cathelicidin LL-37 peptide, are effective against both Gram-negative and Gram-positive infections^[5]. Plant-based secondary metabolites are gaining more attention among natural or synthetic antimicrobial agents. Plants contain biologically active metabolites, known as phytochemicals, that can be extracted from various plant parts, including the leaves, barks, seeds, flowers, and roots, and can be subsequently employed as sources of antimicrobial compounds^[6,7]. The extraction of these compounds can be affected by both the solvent selected for extraction as well as by the extraction method followed. Similarly, the selection of mobile phase solvents for separating phytochemicals from the extract is also very important. Therefore, optimization of extraction method conditions and mobile phase combinations is required to separate and identify these complex plant-based bioactive secondary metabolites from crude extracts of medicinal plants^[8].

Taxus wallichiana (Zucc.) is a significant evergreen tree of medicinal importance, it is being used to extract the medicine taxol from its bark, and because of that it has drawn a lot of attention. Generally, it can be grown in temperate regions of the Indian Himalayan region (IHR) between 1,800 and 3,300 m above sea level^[9]. The Unani and Ayurvedic medical systems have both looked into the medicinal applications of this species. *T. wallichiana* crude extracts also include bioactive metabolites, frequently with potential antioxidant effects. *T. wallichiana* is renowned for its numerous ethno-medical applications^[10]. Asthma and bronchial problems are managed with its needle paste^[11]. Additionally, Himalayan tribal people use tea made from the needles, stem, and bark of Himalayan yew to treat colds, coughs, and hypertension. It has been found that lignan compounds from *T. baccata* heartwood exhibit both antibacterial and cytotoxic properties against a panel of oncology cell lines^[12]. *T. wallichiana* plant part extracts have been researched mainly for their potential for different types of cancer treatment and with little focus on their antibacterial activities^[10,13]. The antimicrobial potential of the *Taxus wallichiana* plant's needles, bark, and stem have been reported, and according to qualitative assessments, actinobacteria, fungi, and bacteria (both Gram-positive and Gram-negative) were all inhibited by the crude extracts of the plant. The initial findings of the quantitative estimations made to the minimum inhibitory concentration also provided support for it^[14–16]. A detailed

study is required for the identification of antimicrobial compounds present in *T. wallichiana* plant part extracts.

The goal of the current work is to separate and collect the fractions of *T. wallichiana*- plant parts extracts for screening and detection of antimicrobial compounds and then characterize the antimicrobial compounds for their identification to better understand their biochemical makeup.

Material and methods

Study site and sample collection

Plant samples (needle, bark, and stem) were collected from the Jageshwar area, District Almora (29°35'–29°39' N and 79°59'–79°53' E) of Uttarakhand, India. A collection of plant samples were submitted to the herbarium records of G.B. Pant National Institute of Himalayan Environment, Kosi-Katarmal, Almora, Uttarakhand, India (Voucher number: GBPI 5050), and the plant was identified as an evergreen woody tree of *Taxaceae* family, it was identified by Dr. K. Chandra Sekar working as Scientist 'F' at the Institute. The gathered plant samples were ground into a fine powder for the next experiments after being cleaned and allowed to air dry.

Chemicals and solvents

HPLC-grade ethanol, methanol, acetone, chloroform, ethyl acetate (E. acetate), hexane, and acetic acid were aquired from Merck, India. Additionally, tryptone yeast extract broth, potato dextrose broth, agar, formic acid, benzene, and toluene were obtained from Hi-media, India. Furthermore, arachidic acid, behenic acid, palmitic acid, stearic acid, myo-inositol, hexadecane, timolol, nicotinamide, procainamide, and cinchonine were purchased from Sigma Aldrich, India. Lastly, the silica gel (mesh 60–120) and TLC plates (Silica gel 60 F254) were procured from Merck Germany.

Microorganisms

Two Gram-positive bacteria, *Bacillus subtilis* (NRRL B30408) and *B. megaterium* (MCC3124). Two Gram-negative bacteria, *Escherichia coli*, and *Serratia marcescens* (MTCC4822). Two Actinobacteria, *Nocardia tenifensis* (MCC2012), and *Streptomyces* sp. (MCC2003). Five fungi, i.e., *Paecilomyces variotii* (ITCC3710), *Aspergillus niger* (ITCC2546), *Fusarium oxysporum* (ITCC4219), *F. solani* (ITCC 5017), and *Trametes hirsuta* (MTCC11397) were used in this study. The microbiology lab of the Institute G.B. Pant National Institute of Himalayan Environment in Almora, Uttarakhand, constructed a microbial culture collection from which these test microorganisms were collected.

Extraction using the maceration method

Two grams of plant material (needle, bark, and stem individually) were combined in a 1:5 ratio (dry powder : solvents) with methanol, ethanol, and ethyl acetate. Using parafilm, the conical flask's mouth was sealed. Using a Remi rotary shaker set to 160 rpm for 48 h at room temperature, samples were macerated. Each solvent extract was vacuum-dried at 35–40 °C using a vacuum oven (Narang Scientific Works, New Delhi, India, Model-257) following extraction. The dried extract was then dissolved in 2 mL of the appropriate solvent, kept apart, and used later at 4 °C.

Qualitative and quantitative phytochemical analysis

Various qualitative tests were performed for phytochemical screening following the method described by Gul et al.^[17]. All

the plant parts extracts were evaluated for total phenolic content, total flavonoid content, and total tannin content, with slight modification (concerning extract volume) in methods as described by Kumaran & Karunakaran^[18], Quettier et al.^[19], and Nwinuka et al.^[20], respectively. The results of total phenolic content are expressed in terms of mg gallic acid equivalent per g dry weight of extract (GAE mg/g dw), flavonoid concentration in terms of mg quercetin equivalent per g dry weight of extract (QE mg/g dw), and tannin content in terms of mg tannic acid equivalent per g dry weight of extract (mg/g dw). The standard equations derived from the calibration curves were utilized for quantification.

Preliminary screening of antimicrobial activity in different plant parts i.e., stem needles, and bark

Agar plate-based bioassays employing the disc diffusion method were carried out to estimate the antibacterial potential of *T. wallichiana* plant part extracts qualitatively. Fungal culture suspensions were made in Potato Dextrose (PD) agar, whilst bacterial and actinobacterial culture suspensions were made on Tryptone Yeast extract (TYE) agar. With the use of a glass spreader, 100 µL of each test organism (separately) was evenly distributed on the corresponding agar surface (TYE agar plates for bacteria and actinobacteria, and PD agar plates for fungus). With the use of sterile forceps, sterile 5 mm filter paper (Whatman No. 1) discs were put over the agar surface. The agar disc was covered with 15 µL of extract. After that, the plates were incubated at 25 °C. Based on the zone of inhibition (mm) measurements after 24 h for bacteria and 120 h for actinobacteria and fungus, the findings were recorded. Each experiment was carried out in triplicate.

Determination of antimicrobial compounds in the crude needle extracts

Sequential extraction for TLC bio-autography

Ten grams of powdered dried needles were macerated in a 1:5 (w/v) mixture of hexane, chloroform, ethyl acetate, methanol and water. Hexane, chloroform, ethyl acetate, methanol, and water were used in increasing order of solvent polarity during the maceration of the same sample. Each solvent extract was vacuum dried after extraction using a vacuum oven (Narang Scientific Works, New-Delhi, India, Model-257) at 35 to 40 °C. The dried extract was then individually dissolved in 2 mL of each solvent.

Separation and identification of antimicrobial compounds

Column and thin layer chromatography (TLC) extracts with antimicrobial activity were further used to isolate and identify antimicrobial compounds. The selectivity triangle approach, created by Snyder, was used to optimize the mobile phase for the chromatographic apparatus^[21]. Through column chromatography, extracts with antibacterial activity were purified using a glass column (32 cm) filled with silica gel (60–120 mesh). Hexane was used to charge the silica bed before a 1 mL sample was put into the column and 10 different mobile phase combinations listed in Table 1 were used to elute the sample. After collecting the eluted fractions, TLC-bioautography was used to further separate the fractions that have antibacterial activity.

Based on the selectivity triangle, a TLC plate (silica gel 60 F254, Merck, Germany) was used to separate the fraction with

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antibacterial activity using various combinations of mobile phases (Table 2). The TLC plates were examined for antimicrobial activity, and any areas with potential for antimicrobial activity were scraped off and dissolved in ethyl acetate. Finally, utilizing the solvent systems listed in Table 3, column chromatography was performed once more to further purify fractions to isolate molecules responsible for antibacterial activity. Afterward, the above-mentioned procedure for testing the antimicrobial activity was used for testing the antimicrobial properties of the fractions and was evaluated using a qualitative test (the disc diffusion assay) and a quantitative assessment (the minimum inhibitory concentration, or MIC). The same mobile phase from which the fractions were taken was utilized as a control.

Determination of antimicrobial compounds in the selected rich fractions

Gas chromatography and mass spectrometry (GC/MS)

With the use of the Shimadzu GC/MS-QP2010 ultra, certain fractions were examined. The eluted components' electron impact (EI) mass spectrum was captured using a 10 μ L injection

volume. The column's (DB35-MS Capillary non-polar column with dimensions of 30 mm \times 0.25 mm ID \times 0.25 μ m film) initial temperature was maintained at 50 $^{\circ}$ C for 2 min, after which it was raised to 250 $^{\circ}$ C at a rate of 7 $^{\circ}$ C/min for a hold time of 3 min, and then to 280 $^{\circ}$ C for 18 min at a rate of 10 $^{\circ}$ C/min. The carrier gas (Helium) flow rate was held constant at 1.21 mL/min. The mass range of 50–1,000 Da has been scanned. NIST 14 and Wiley 8 library spectra tools were used to identify the chemical components.

Liquid chromatography and mass spectrometry (LC/MS)

Using LC/MS (Waters, MS Synaptic GZ HDMS LC-MS UPLC H-Class) with a C18 (1.7 μ m) column, injection volume of 20 μ L, and 0.1% formic acid: acetonitrile: methanol (20:30:50 v/v/v; isocratic mode) as mobile phase, the non-volatile bioactive chemicals included in the extract were identified. The column was kept at a constant temperature of 35 $^{\circ}$ C. The chromatogram and mass spectrum analysis were done and the results were compared to the reference mass spectrum found in the Metlin library and the online tool Massbank.

FTIR (Fourier Transform Infrared Spectroscopy)

The Agilent-Cary 630 FT-IR spectrometer was used to record the FTIR spectra. Using Microlab FTIR software, 20–40 μ L of samples were used for FTIR analysis spanning the frequency range of 650–4,000 cm^{-1} at 8 cm^{-1} intervals. Based on the wave number, the spectrum was further examined for the presence of potential functional groups.

Antimicrobial effects of recognized compounds and their detection in needle extract

Following the identification of antimicrobial compounds, all compounds were procured from Sigma, India, and their antimicrobial activity was assessed through the utilization of the technique delineated in an earlier section. Subsequently, the fatty acids in the needle methanolic extracts are measured. Gas chromatography (Chemito GC, Ceres 800 plus) with a Flame Ionization Detector (FID) was used to evaluate the fatty acids. A BP20 column (30 m \times 0.25 mm) in splitless mode was used for the analysis, and 10 μ L of the extract was injected. The stationary phase film thickness was 0.25 mm. The temperatures of the injector and detector were adjusted to 250 and 290 $^{\circ}$ C, respectively. The flow rate of the carrier gas, helium, was kept constant at 0.75 mL/min. The oven was preheated at 60 $^{\circ}$ C, then raised to 260 $^{\circ}$ C at a pace of 3 $^{\circ}$ C per min, and left there for 10 min. By comparing the retention durations of the fatty acids to a variety of fatty acid standards, the fatty acids were found.

Using RP-HPLC (Shimadzu LC solution, Japan) and a PDA (Photodiode Array) detector, non-volatile chemicals were examined. Using a C18 reverse-phase column, the active

Table 1. Solvent strength of mobile phase used for separation of compounds using column chromatography.

| Solvent strength | | |
|-----------------------------------|-----------------|------|
| Mobile phase I (%) | | |
| Hexane (100) | – | 0 |
| Hexane (90) | E. acetate (10) | 0.44 |
| Hexane (80) | E. acetate (20) | 0.88 |
| Hexane (70) | E. acetate (30) | 1.32 |
| Hexane (60) | E. acetate (40) | 1.76 |
| Hexane (50) | E. acetate (50) | 2.2 |
| Hexane (40) | E. acetate (60) | 2.64 |
| Hexane (30) | E. acetate (70) | 3.08 |
| Hexane (20) | E. acetate (80) | 3.52 |
| Hexane (10) | E. acetate (90) | 3.96 |
| Mobile phase II (%) | | |
| E. acetate (100) | – | 4.4 |
| E. acetate (90) | Methanol (10) | 4.47 |
| E. acetate (80) | Methanol (20) | 4.54 |
| E. acetate (70) | Methanol (30) | 4.61 |
| E. acetate (60) | Methanol (40) | 4.68 |
| E. acetate (50) | Methanol (50) | 4.75 |
| E. acetate (40) | Methanol (60) | 4.82 |
| E. acetate (30) | Methanol (70) | 4.89 |
| E. acetate (20) | Methanol (80) | 4.96 |
| E. acetate (10) | Methanol (90) | 5.03 |
| Mobile phase III (%) | | |
| Acetic acid (0.1) Methanol (99.9) | 5.1 | |

Table 2. Solvent strength of mobile phase used for separation of compounds using thin layer chromatography.

| Code | Mobile phase | Ratio | Solvent strength | Separation | Spot | Rf value |
|------|---|-----------|------------------|------------|------|--|
| M1 | Chloroform : E. acetate : Formic acid | 5:4:1 | 4.4 | Yes | 2 | 0.6, 0.7 |
| M2 | E. acetate : Methanol : Benzene | 2:0.5:2.5 | 3.6 | Yes | 3 | 0.3, 0.6, 0.68 |
| M3 | Ethanol : Chloroform | 1:1 | 4.2 | Yes | 4 | 0.4, 0.56, 0.64, 0.83 |
| M4 | Toluene : E. acetate : Formic acid | 6:4:0.5 | 3.2 | Yes | 5 | 0.25, 0.38, 0.43, 0.64 and 0.81 |
| M5 | Hexane : Acetone : Toluene : Ethanol | 10:7:7:6 | 5.2 | No | No | No |
| M6 | Chloroform : Acetonitrile | 7:3 | 4.6 | Yes | 3 | 0.77, 0.8, 0.85 |
| M7 | Chloroform : Methanol | 7:1 | 3.9 | Yes | 8 | 0.67, 0.15, 0.19, 0.32, 0.40, 0.45, 0.62, 0.75 |
| M8 | E. acetate : 2-Propanol | 95:5 | 4.3 | No | No | No |
| M9 | Chloroform : E. acetate : Methanol | 25:20:5 | 4.3 | Yes | 5 | 0.25, 0.4, 0.54, 0.68, 0.76 |
| M10 | Chloroform : Formic acid : Acetonitrile | 6:0.5:3.5 | 3.3 | No | No | No |

Table 3. Solvent strength of mobile phase used for separation of compounds present in spot 2 through thin layer chromatography and column chromatography.

| Code | Mobile phase | Ratio | Solvent strength |
|------|--------------------------------|-----------|------------------|
| M11 | Hexane : E. acetate | 7:3 | 1.32 |
| M12 | Toluene : Methanol | 8:2 | 2.94 |
| M13 | Methanol : Chloroform | 1:1 | 4.6 |
| M14 | DMF : DCM : Acetonitrile | 5:1:4 | 5.83 |
| M15 | Acetic acid : Methanol : Water | 5:2.5:2.5 | 6.77 |

metabolites in methanolic needle extracts were measured. Every sample had an injection volume of 20 µL, and the mobile phase was operated in gradient mode. The following compounds' wavelengths and mobile phase compositions were measured: procainamide (275 nm, methanol : water), nicotinamide (275 nm, 0.1 % TFA in water : acetonitrile : methanol), myoinositol (280 nm, methanol : 0.02 M H₂SO₄), and cinchonine (260 nm, acetonitrile : methanol : phosphoric acid), with a flow rate of 1 mL/min for all the compounds, except 1.2 mL/min for nicotinamide, and 0.6 mL/min for myoinositol.

Statistical analysis

Data were collected from all trials in triplicate and were expressed as a mean with standard deviation (SD). GraphPad, Prisma 8 software was used to create the heatmap.

Results and discussion

Phytochemical analysis

The extract yield was higher in the methanolic extract (15%) followed by ethanol (12%) and ethyl acetate (9%). The qualitative analysis of *T. wallichiana* needle, stem, and bark shows the presence of phytochemical substances such as phenol, flavanol, flavonoid, tannin, saponin, terpenoid, glycoside, and steroid (Fig. 1). Total phenolic, flavonoids and tannin content were higher in ethanolic extracts of needles (82.20 ± 0.51 mg GAE/g (dw), 79.72 ± 0.69 mg QRE/g (dw), 12.98 ± 0.34 mg TAE/g (dw)) and methanol (81.37 ± 0.85 mg GAE/g (dw), 54.10 ± 0.58 mg QRE/g (dw), 10.52 ± 0.29 mg TAE/g (dw)) extracts, respectively. In stem, phenolic (52.57 ± 0.39 mg GAE/g (dw)) and tannin (9.09 ± 0.11 mg TAE/g (dw)) content was higher in methanolic extracts while flavonoids content was higher in ethanolic extracts (38.41±0.68 mg QRE/g (dw)). Likewise for bark, phenolic (68.52 ± 0.72 mg GAE/g (dw)) content, flavonoids (61.57 ± 1.09 mg QRE/g (dw)), and tannin (13.05 ± 0.18 mg TAE/g (dw)) content were higher in ethyl acetate extract. Results are shown in Fig. 2.

Antimicrobial potential of needle, bark, and stem

Using extracts from *T. wallichiana* needle, bark, and stem a preliminary test was conducted to determine which plant part would be most effective for the isolation and identification of antimicrobial metabolites. It's interesting to note that all plant part extracts exhibited antimicrobial action against the three types of microorganisms i.e., fungi, actinobacteria, and bacteria. *T. wallichiana* extracts; were found more effective against *Bacillus* species in terms of inhibition, and were less effective against *E. coli*. Between two tested *Bacillus* species, ethanol needle extract demonstrated greater inhibition for *B. subtilis*, whereas needle methanol extract demonstrated greater inhibition for *B. megaterium*. Likewise, needle and bark methanolic extracts showed higher antimicrobial activity for *S. marcescens*, needle and bark ethanolic extracts for *P. chlororaphis*, and needle methanol and ethanol extract for *E. coli*. Methanolic extracts of needles and bark showed significant inhibition against actinobacteria also i.e., *N. tenirefensis*, and *Streptomyces* sp., Fig. 3 displays the zone of inhibition (mm) results. *T. wallichiana* parts extracts showed antifungal activity against five fungus species i.e., *P. variotii*, *T. hirsuta*, *F. oxysporum*, *A. niger*, *F. solani*, and *P. variotii* showed the highest zone of inhibition with needle ethanolic extracts, *T. hirsuta* with bark ethanolic extract, *F. oxysporum*, and *A. niger* with needle methanol, and *F. solani* with needle and bark ethyl acetate extracts. Overall, the extent of the zone of inhibition for ethanolic and methanolic plant component extracts demonstrated strong antibacterial activity (Fig. 3).

Screening for the best extracts for further purification studies

Several combinations of mobile phases were tried using the selectivity triangle approach for TLC, column chromatography, and bioautography to separate antimicrobial chemicals in needle fractions. Column chromatography and TLC have been used to identify and further purify a large number of bioactive fractions. The most often used techniques are thin-layer chromatography (TLC) and column chromatography because they are affordable, convenient, and available in a variety of stationary phases^[22].

Based on the results of the preliminary test of phytochemical and antimicrobial assay, needles were selected for further studies of antimicrobial compound isolation and characterization. There are several bioactive secondary metabolites from various classes included in the plant's crude organic extracts. When extracting desired compounds, it is beneficial to use multiple solvents sequentially, starting from those with low polarity and gradually increasing to high polarity. This

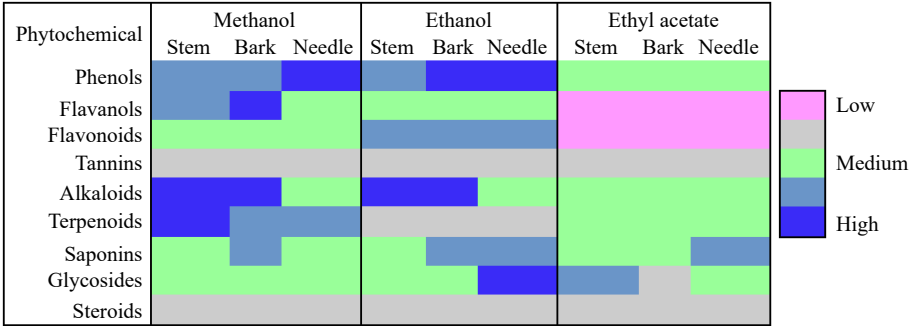


Fig. 1 Qualitative phytochemical screening for stem, bark, and needle extracts.

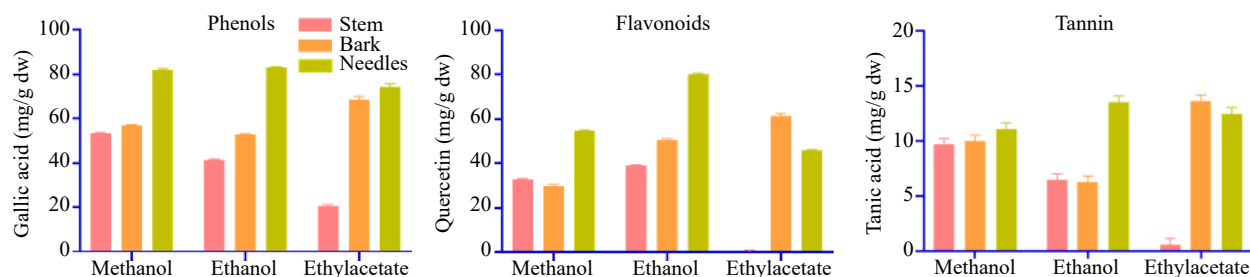
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Fig. 2 Total phenolic content, total flavonoids content, and total tannin content of stem, bark, and needle extracts of *T. wallichiana*.

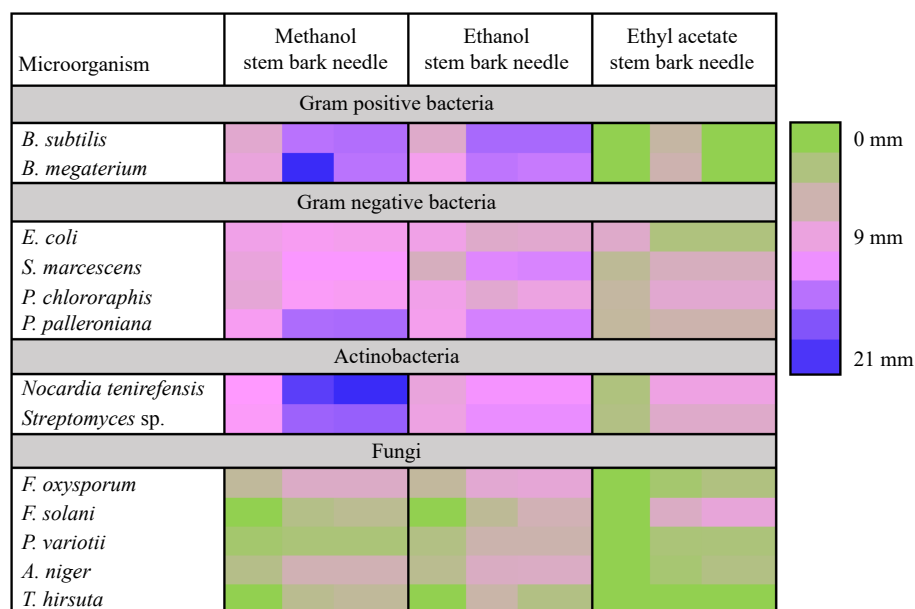


Fig. 3 Heatmap showing preliminary screening of antimicrobial potential (zone of inhibition (mm)) of needle, stem, and bark of *T. wallichiana*.

approach helps prevent overlapping similar types of secondary metabolites in the desired extracts. Based on that in the current study, utilized hexane, chloroform, ethyl acetate, methanol, and water were utilized. It was found that only the extracts from methanol and ethyl acetate (ME & EAE) displayed antimicrobial activity, while the hexane, chloroform, and aqueous extracts exhibited no such activity.

Based on the results of antimicrobial activity ME & EAE were selected for further fractionation using TLC and column chromatography. It is essential to fully understand the nature of the existing secondary metabolites in the extract before developing the mobile phase for column and thin-layer chromatography. In this context, Gas Chromatography-Mass Spectrometry (GC-MS) was conducted for the extracts (ME & EAE) exhibiting antimicrobial activity. The extract's mass spectra revealed the presence of carboxylic sugars, fatty acids, phenols, sterols, and alkanes. Benzene, propanol, gamma-sitosterol, phenols, and benzoic acid were prevalent in both extracts. The findings mentioned have been previously documented in the essential oil of *Nasturtium officinale*^[23] and in the leaf extract of *Adiantum capillus*^[24].

Mobile phase designing for column chromatography

The secondary metabolites identified in the crude extract were then utilized to create the mobile phases for column chromatography, which was utilized to separate the secondary

metabolites further, using a selectivity triangle. In the present study, optimization of the mobile phase was done through a selectivity triangle. It ultimately identifies and quantifies the blend of intermolecular interactions that occur between solutes and solvents/phases^[25]. The basic principle of the selectivity triangle is shown in Fig. 4 and details on the first set of mobile phases are shown in Table 1.

Bioassay-guided fractionation and screening for antimicrobial activity

Ten-bed volumes of mobile phase (details are given in Table 1) were passed through the column containing methanol and ethyl acetate extracts. Collected fractions were subjected to testing for antimicrobial activity. The methanol extract fractions exhibited selective antibacterial activity, while the ethyl acetate extract fractions demonstrated robust antimicrobial and antifungal activity. Following the results, the ethyl acetate extract fractions were chosen for further bioautography. A schematic diagram explaining the results is shown in Fig. 5.

TLC-bioautography

Isolated fractions with antimicrobial activity were further subjected to separation of antimicrobial compounds using a bioautographic technique^[26–28]. With the help of a selectivity triangle, 10 different mobile phases were designed for the separation of compounds through TLC (Table 2). From the 10 mobile phases, three did not show any separation. The mobile

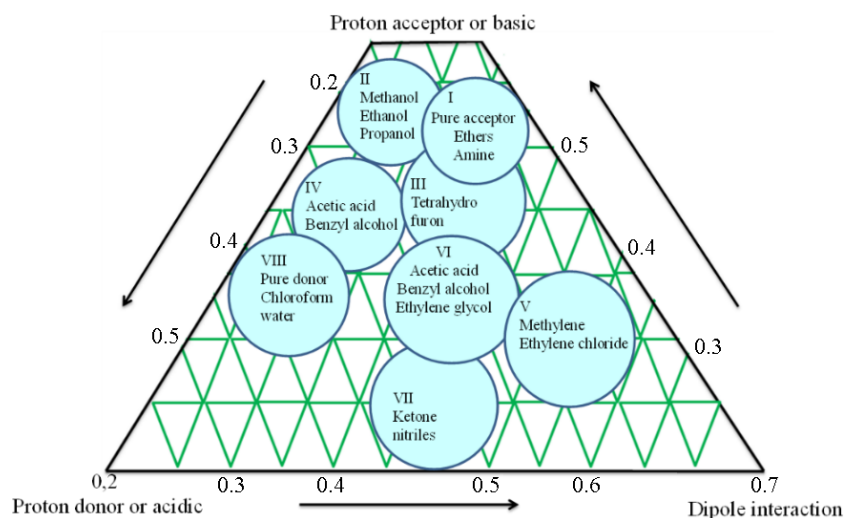


Fig. 4 Schematic diagram of selectivity triangle principle for mobile phase optimization.

phases, which had shown an RF value between 0.2–0.8 cm, were considered as good (Table 2).

Seven mobile phases M1, M2, M3, M4, M6, M7, and M9 (Table 2) were found suitable for the separation of compounds present in the fraction having antimicrobial activity. Based on Rf values, mobile phase M3 and M4 were used for further study. During bioautography out of four spots, three spots showed antimicrobial activity. Spot 1, 2, and 3 have antimicrobial activity. Spot 1 and 3 have selective antibacterial activity and spot 2 has antibacterial, antiactinobacterial, and antifungal activity. Therefore, spot 2 was selected for further separations and detection using column chromatography and thin-layer chromatography. However, spot 2 did not separate through M3 and M4. Hence, another set of mobile phases was designed with the help of a selectivity triangle to improve the separation process. The solvent strength of the new mobile phase is given in Table 3. The schematic diagram is shown in Fig. 6a.

Screening for the best fractions having potential antimicrobial activity

Ten-bed volumes were passed through the column packed with spot 2. Fractions were collected through six different mobile phases, and afterward, all the fraction was then tested

for their antimicrobial activity. After this exercise, three rich fractions (FA, FB, and FC) were selected based on their antimicrobial activity against all three tested groups including bacteria, actinobacteria, and fungi. The schematic diagram is shown in Fig. 6b and the results on the zone of inhibition are shown in the heatmap of Fig. 7a.

Description of antimicrobial compounds

All of the fractions were subjected to the identification and characterization of antimicrobial compounds after the final three rich fractions (FA, FB, and FC) with antibacterial activity were chosen. Rich fractions (FA, FB, and FC) were subjected to GCMS (for volatile chemicals), LCMS (for non-volatile compounds), and FTIR (for functional group) analyses to characterize antimicrobial substances. A list of identified compounds through LCMS and GCMS are given in Table 4. Compounds like benzoic acid, hexadecenoic acid, or palmitic acid were found in all three fractions according to GCMS analysis. Fatty acids, aromatic carboxylic acids, aliphatic hydrocarbons, and sesquiterpenoid alcohols made up the bulk of the compounds that were found. Likewise, the existence of many categories of substances, such as alkaloids, vitamins, flavones, quinones, carboxylic acid, lipids, etc., was revealed by LCMS analysis.

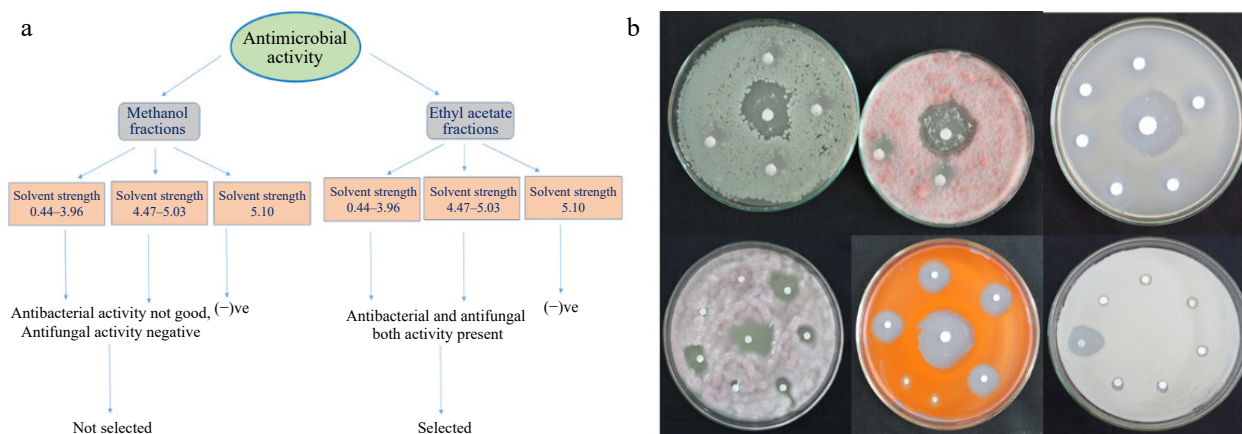


Fig. 5 (a) Schematic representation of different subfractions collected from fractionation of methanol and ethyl acetate fraction of *T. wallichiana* needle. (b) Antimicrobial activity of collected fractions of ethyl acetate extract through column chromatography.

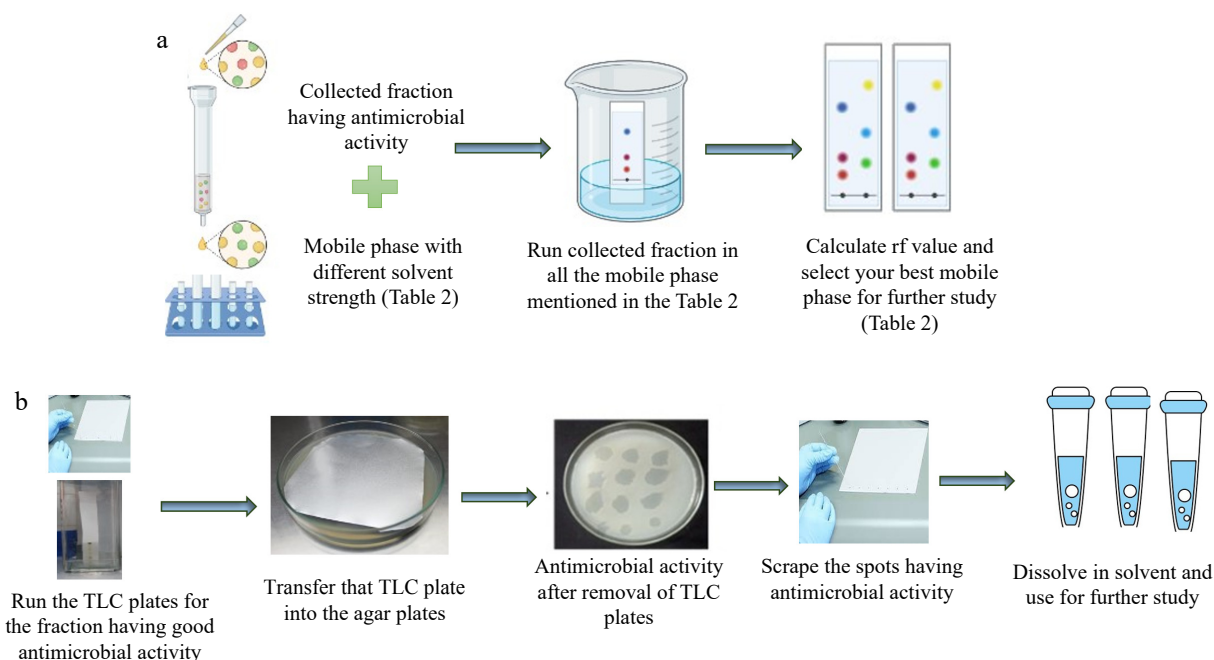
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Fig. 6 (a) Schematic diagram for finalizing the mobile phase for the TLC. (b) Schematic diagram for the TLC-bioautography.

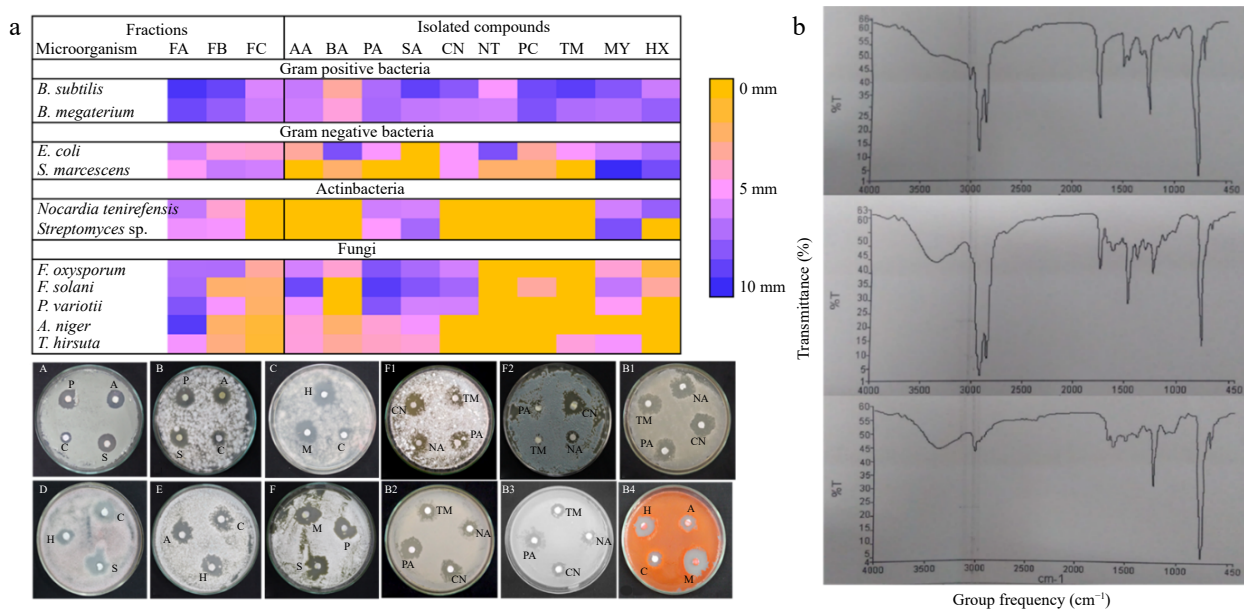


Fig. 7 (a) Heatmap showing antimicrobial potential (zone of inhibition (mm)) of final fraction and isolated compounds having antimicrobial activity. PA= palmitic acid, SA= stearic acid, AA= arachidic acid, MY= myoinositol, and HA= hexadecane, PA = procainamide, CN = cinchonine, NA = nicotinamide, TM = timolol. B1) *B. megaterium*, B2) *B. Subtilis*, B3) *E. coli*; and B4) *S. marcescens*. (A and F2) *A. niger*, B and C *P. variotii*, D) *F. oxysporum* E and F) *T. hirsuta*. B) FTIR spectra of rich fraction FA, FB, and FC of *T. wallichiana* needles.

To find out what functional groups were present in the identified compounds in the purified fraction, FTIR analysis was performed. Based on the fingerprint region ($1,500\text{--}4,000\text{ cm}^{-1}$) and functional groups, FTIR spectra were analyzed. Figure 7b displays the FTIR spectrum. The functional groups of the compounds identified by GC-MS and LC-MS were compared with the FTIR spectra. All of the fractions share bands between $1,000$ and $1,275\text{ cm}^{-1}$ (C-O/C-N str) between $3,000$ and $3,100\text{ cm}^{-1}$ (C-H str in sp^2 hybridized carbon, such as alkene = C-H str or aromatic C-H str). The bands at $3,362\text{ cm}^{-1}$, which are

common between FB and FC, and the bands between $2,800$ and $3,000\text{ cm}^{-1}$ (C-H str in sp^3 hybridized carbon, such as methyl and methylene C-H str), which were common between FA and FB, suggested that N-H stretching may have occurred based on the band's shape, while broadness suggested the possibility of a carboxylic group or ester. Additionally, the tiny hump at $3,362\text{ cm}^{-1}$ suggested that FB and FC may include secondary or tertiary amides. In FA and FB, the band about $2,850$ suggested the presence of methyl and methylene C-H str; in fraction FC, this was less evident. The only bands found in FC

Table 4. A list of identified compounds in fractions showed antimicrobial activity.

| Fraction (FA) compounds | Fraction (FB) compounds | Fraction (FC) compounds |
|---------------------------------------|--------------------------------|----------------------------|
| GC-MS | | |
| 1,6-Octadine-3-ol | Benzoic acid | Benzenepropanol |
| Benzoic acid | 4-Tetradecene | Megastigmatrienone |
| Benzene, 1-methoxy 4-(2-propenyl) | 2,4-Ditert-butylphenol | Ar-tumerone |
| Phenol, 2-methyl-5-(1-methylethyl) | 9-Octadecanoic acid | Mome-inositol |
| 1-Tridecene | Hexadecane | Hexadecanoic acid |
| Pentadecane | 1,4-Dimethyl-2 phenoxybenzene | 9-Octadecenoic acid |
| Succinic acid | E-15-Heptadecenal | Eicosanoic acid |
| E-14-Hexadecenal | Benzene dicarboxylic acid | Di-n-octyl phthalate |
| Neophytadiene | Hexadecanoic acid | |
| Hexadecanoic acid | 1-Nonadecene | |
| 1-Nonadecene | Ecosanoic acid | |
| Docosanoic acid | Octacosanol | |
| 1-Tetradecanol | Phenol, 2,4-bis(1-phenylethyl) | |
| Benzene dicarboxylic acid | Di-n-octyl phthalate | |
| LC-MS | | |
| 1-aminocyclopropane-1-carboxylic acid | Nicotinamide | Nicotinamide |
| Methyl methanethiosulfonate | Cinchonine | Dodecylsulfonilacetic acid |
| Methionyl-Glycin | Ranitidine | Diphenoxylac acid |
| Ergothioneine | Psoralenol | Cinchonine |
| Procainamide | Squamocin B | Ergothioneine |
| Alcoifosfamide | Trimethaphan | |
| Timolol | Timolol | |
| Boviquinone | Trimethaphan | |
| Lobaric acid | Frangulanine | |

were around 3,017 cm^{-1} and the band at 1,675.95 cm^{-1} , which are typically caused by the alkenyl C=C stretch/amide and the aromatic C-H stretch, respectively.

Antimicrobial compounds in needle-rich fractions (FA, FB, and FC)

After all the biochemical characterization i.e., GC-MS, LC-MS, and FTIR, antimicrobial compounds were identified in all three fractions (Table 5). Ten compounds i.e., arachidic acid, behenic acid, palmitic acid, and stearic acid (fatty acid); vitamins (nicotinamide); alkaloids (cinchonine, timolol); amino benzamides (procainamide); carbocyclic sugars (myoinositol); and alkane hydrocarbons (hexadecane) were identified among all the compounds in the three rich fractions (FA, FB, and FC) that had antibacterial activity. To assess the antibacterial activity of the standard compounds against all the specified microorganisms, plate-based bioassays and minimum inhibitory concentration (MIC) were employed. Results in detail are given in Fig. 7a. Identified fatty acids were found potential antibacterial and

antifungal agents. Likewise in some previous reports also using bioassay-guided fractionation, antibacterial fatty acids have been extracted from a variety of plants. Yff et al.^[29] isolated palmitic acid from *Pentania prunelloides* which was effective towards bacteriological contagions. Palmitic and stearic acids are reported in *Labisia pumila* leaves with antibacterial activities^[30,31].

Shafaghath^[32] has reported the antibacterial potential of fatty acids derived from the various parts of the plant *Hypericum scabrum* (flower, leaf, stem, and seed). Cerdeiras et al.^[33] reported the antibacterial activity of stearic acid, present in the aerial sections of *Ibicella lutea*. In addition to fatty acids, six more compounds i.e., myoinositol, hexadecane, chinchonine, timolol, procainamide, and nicotinamide were present as per MS data and these compounds also showed antibacterial properties along with antifungal activities. These substances have specific antifungal action in addition to strong antibacterial activity. Comparably, reports of myoinositol's antibacterial properties

Table 5. List of compounds identified in the separated fraction (spot 2) having antimicrobial potential.

| S. No. | Compounds | Formula | Molar mass (g/mol) | Classification | Concentration in needles (mg/g (dw)) |
|----------------------|----------------|---|--------------------|----------------------|--------------------------------------|
| GC-FID analysis | | | | | |
| 1 | Arachidic acid | C ₂₀ H ₄₀ O ₂ | 312.53 | Saturated fatty acid | 22.94 ± 0.09 |
| 2 | Behenic acid | C ₂₂ H ₄₄ O ₂ | 340.58 | Saturated fatty acid | 31.04 ± 0.05 |
| 3 | Palmitic acid | C ₁₆ H ₃₂ O ₂ | 256.43 | Saturated fatty acid | 16.81 ± 0.03 |
| 4 | Stearic acid | C ₁₈ H ₃₆ O ₂ | 284.48 | Saturated fatty acid | 20.10 ± 0.06 |
| RP-HPLC-PDA analysis | | | | | |
| 5 | Cinchonine | C ₁₉ H ₂₂ N ₂ O | 294.17 | Alkaloid | 3.75 ± 1.21 |
| 6 | Nicotinamide | C ₆ H ₆ N ₂ O | 122.12 | Vitamin | 21.14 ± 0.53 |
| 7 | Procainamide | C ₁₃ H ₂₁ N ₃ O | 235.325 | aminobenzamides | 14.61 ± 0.71 |
| 8 | Timolol | C ₁₃ H ₂₄ N ₄ O ₃ S | 316.421 | Alkaloid | 6.31 ± 0.54 |
| 9 | Myoinositol | C ₆ H ₁₂ O ₆ | 180.16 | carbocyclic sugar | 1.45 ± 1.01 |
| 10 | Hexadecane | C ₁₆ H ₃₄ | 226.41 | alkane hydrocarbon | – |

Detecting antimicrobial compounds from *Taxus wallichiana*

have been made^[34,35]. Hexadecane from *Allium nigrum* has also been shown to have antibacterial activity^[36]. Finally, through this piece of work, a separate fraction of *T. wallichiana* needle extract was obtained with known compounds, which have shown good antimicrobial activity with selected microbial species. This can be further utilized for in-depth studies with other types of microbes. After testing for antimicrobial activity, the compounds in the needle extracts were quantified using GC and HPLC alongside their standard compounds. The results of the quantification are given in Table 5.

Conclusions

Due to the increasing demand for *T. wallichiana* bark, there has been a continuous decline in the availability of raw materials sourced from its natural habitats. Consequently, there is a notable scarcity of information concerning various aspects such as the phytochemical composition and antimicrobial properties of *T. wallichiana*. In the current research, a thorough examination was conducted, emphasizing the diversity of phytochemicals, antimicrobial functions, and selection of suitable solvents for the extraction of bioactive components responsible for antimicrobial activity. Needles showed the highest antimicrobial potential among the stem, bark, and needles of *T. wallichiana*. Based on the preliminary data, rich-fractions of needles were subjected to LC-MS, GC-MS, and FTIR analyses, and a total of 10 compounds with antimicrobial potential i.e., myoinositol, hexadecane, cinchonine, procainamide, nicotinamide, palmitic acid, stearic acid, arachidic acid, behenic acid, and timolol have been identified from the needles. It is also suggested that focusing on utilizing the needles, rather than the bark and stem is optimal for maximizing their antimicrobial potential. Notably, *T. wallichiana*, an evergreen tree, has needles with a lifespan of approximately three and a half months, underscoring the importance of promoting needle usage for sustainable extraction of taxol and other potential biologically active secondary metabolites.

Author contributions

The authors confirm contribution to the paper as follows: study conception and design: Pandey A, Agnihotri V; experiments conduction, data interpretation, draft manuscript writing: Adhikari P; manuscript editing & finalization: Pandey A, Agnihotri V. All authors reviewed the results and approved the final version of the manuscript.

Data availability

The data that support the findings of this study are available on request from the corresponding author.

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Conflict of interest

The authors declare that they have no conflict of interest.

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