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Effects of substituting mineral nitrogen fertilizer with organic manure on nitrogen transformations and nitrous oxide emissions from tobacco (*Nicotiana tabacum* L.) fields

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Abstract

The partial substitution of mineral nitrogen fertilizer (SN) with organic manure, hereafter termed organic substitution, is increasingly regarded as an environmentally preferable and more sustainable nutrient management strategy. In the present research, we systematically investigated how three distinct organic substitution treatments—application of commercial organic manure (NC), traditional farmyard manure (NF), and a bio-organic manure inoculated with *Trichoderma viride* Pers. (NT)—influence gross nitrogen transformation processes and the pathways of nitrous oxide (N₂O) production in *Nicotiana tabacum* L. (tobacco) fields, using plots receiving only SN as the reference treatment. Application of organic substitution treatments markedly enhanced soil nitrogen transformation processes. Relative to SN, these treatments stimulated nitrate (NO₃⁻) production (6.43–7.43 mg N kg⁻¹ soil d⁻¹), increased ammonium (NH₄⁺) production (1.01–1.37 mg N kg⁻¹ soil d⁻¹), and elevated NH₄⁺ consumption (6.50–6.96 mg N kg⁻¹ soil d⁻¹). Autotrophic nitrification dominated NO₃⁻ production and NH₄⁺ consumption, and was markedly enhanced under NF and NT treatments. Although organic substitution treatments increased nitrogen transformations and thereby reduced N₂O emissions from autotrophic nitrification, autotrophic nitrification remained the dominant source of N₂O. By modulating nitrogen transformation processes, organic substitution treatments significantly reduced overall N₂O emissions, providing a basis for sustainable agricultural development and greenhouse gas mitigation.

Keywords: Nicotiana, Organic manure, Gross nitrogen transformation, Nitrous oxide

Highlights

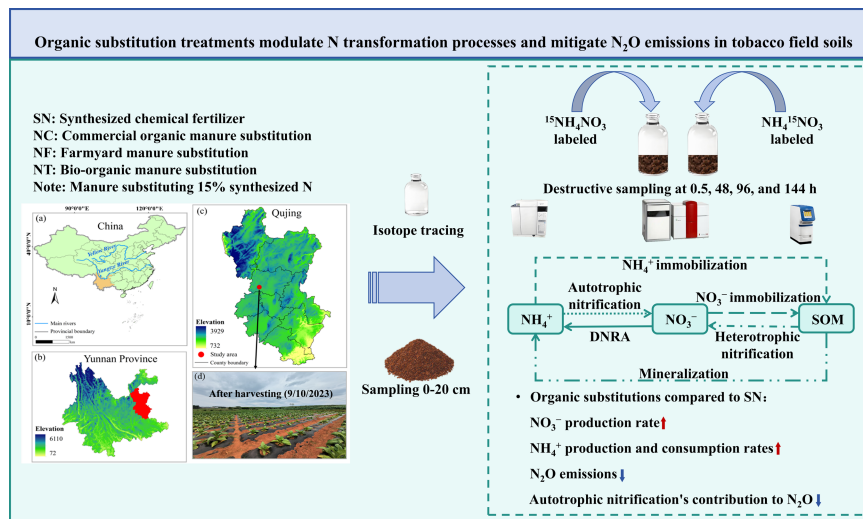
- Organic substitutions increased NO₃⁻ production and NH₄⁺ production/consumption rates.
- Autotrophic nitrification dominated NO₃⁻ production and NH₄⁺ consumption.
- Organic substitutions reduced cumulative N₂O emissions compared to mineral fertilizer.

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Graphical abstract



Introduction

Since the 1980s, Chinese farmers have applied high nitrogen (N) fertilizer rates to overcome N limitation, but this has led to excessive N losses and increased nitrous oxide (N₂O) emissions. With a 100-year global warming potential approximately 273-fold greater than that of carbon dioxide, N₂O exacerbates global warming^[1–3]. In this context, a key scientific and practical challenge for modern agriculture is to decrease reliance on mineral N fertilizers while sustaining crop productivity or limiting yield losses to an acceptable level, so as to reconcile food production demands with greenhouse gas mitigation and broader environmental protection objectives. In response, the Chinese Ministry of Agriculture launched the 'Zero Growth of Chemical Fertilizer' program in 2015, aiming to halt the growth of fertilizer use by 2020^[4].

Among the options for reducing N inputs while improving efficiency, organic substitution—partially substituting chemical N fertilizer with organic manure—has been widely regarded as a more environmentally friendly approach^[5]. It not only enhances agroecosystem sustainability but also exerts a pivotal influence on soil N transformations and utilization^[6]. Organic N in manure, unlike N directly supplied by chemically synthesized fertilizers, is gradually released through microbial decomposition and mineralization, effectively maintaining the dynamic balance of soil N^[7]. In nitrification, organic manures boost the activity of ammonia-oxidizing and nitrite-oxidizing microorganisms by increasing soil organic matter and N sources, thereby improving nitrate N (NO₃⁻) production and soil N availability^[8]. Additionally, organic manures enhance soil structure, promote anaerobic microsites, and supply denitrifying microorganisms with carbon (C) and suitable conditions that can increase denitrification rates. While enhanced denitrification may alleviate NO₃⁻ accumulation and leaching, it may simultaneously increase N₂O emissions, raising concerns about the overall climatic impact^[9,10]. Most previous studies have focused on staple crops such as rice and wheat. Here, we shift the focus to tobacco (*Nicotiana tabacum* L.), a crop that is highly sensitive to N supply and has distinctive nutrient-demand characteristics, with the aim of identifying and elucidating its specific patterns of soil N cycling under organic substitution.

Although numerous studies have examined how organic manures influence soil N transformations, these processes in red soils may differ significantly from those in other soil types because of their distinctive physicochemical characteristics. In southern China, red soils are typically low in organic matter and highly acidic, conditions that inhibit nitrification and slow the conversion of ammonium N (NH₄⁺) to NO₃⁻^[11,12]. Additionally, red soils are rich in Fe–Al oxides, which interact with organic matter and NH₄⁺, accelerating N₂O production during denitrification and inhibiting the subsequent reduction of N₂O to N₂, ultimately intensifying N₂O emissions^[13,14]. Poor aeration in these soils also promotes anaerobic microsites that truncate denitrification at the N₂O stage, resulting in higher N₂O emissions and reduced fertilizer N use efficiency^[15].

Tobacco is highly sensitive to both the form and timing of N supply, and both its productivity and leaf quality are closely coupled to N availability in the root zone. In many tobacco-growing regions of China, including Yunnan, tobacco is cultivated predominantly on acidic red soils with low organic matter content and limited buffering capacity^[16]. Under such conditions, inappropriate N management can readily lead to soil acidification, poor structure, unbalanced NH₄⁺/NO₃⁻ ratios, and elevated N₂O emissions, ultimately constraining both N use efficiency and leaf quality^[17]. The substitution of mineral fertilizer with organic manure in tobacco production entails both opportunities and challenges. On the one hand, it can improve soil physicochemical properties and biological activity, thereby enhancing soil quality and sustaining long-term nutrient supply^[18]. On the other hand, the asynchronous release of N from organic manure over time may not coincide with crop N demand, and the increased availability of labile C and N may enhance ammonia volatilization and N₂O emissions^[19]. Organic manures have the potential to partially replace mineral N fertilizers in these systems. By supplying labile C, they can mitigate soil acidification, improve aggregate stability and pore structure, and stimulate microbial activity, thereby reshaping soil C–N coupling and N transformation rates in tobacco fields. Because tobacco roots may differentially prefer NH₄⁺ vs NO₃⁻ depending on growth stage and environmental conditions, tobacco root exudation traits, together with regional cultivation systems, can jointly shape nitrifying and denitrifying communities, ultimately modulating N transformation rates and N₂O emission pathways^[20].

Soil N transformation processes are complex, and different organic substitution treatments exert distinct effects on these transformations. Bio-organic manure, owing to its microbial inoculants and suitable C/N ratio, can slow nitrification, modulate denitrification, and reduce N₂O emissions^[21]. Commercial organic manure, which undergoes industrial processing and fermentation, typically contains stable, plant-available N forms readily absorbed by crops, thereby mitigating soil N leaching and excessive N transformations and reducing N₂O emissions^[22]. Farmyard manure provides a long-term substrate for soil microorganisms due to its diverse, slow-releasing N sources, resulting in relatively low N₂O emissions despite continuous denitrification^[23]. However, it remains unclear how these contrasting organic substitution treatments differentially affect N transformation processes and N₂O emission pathways in acidic red soils under tobacco cultivation, and whether there are specific thresholds or applicability ranges for each treatment in such systems. Given the prevalence of acidic red soils in Yunnan, systematically evaluating organic substitution treatments and identifying their applicability ranges and thresholds are indispensable for enhancing N use efficiency and developing N₂O emission-reduction treatments.

This study was designed to elucidate how distinct organic substitution treatments influence gross N transformation and N₂O emission pathways, and their underlying mechanisms. Using ¹⁵N labeling techniques (e.g., ¹⁵NH₄NO₃ and NH₄⁺¹⁵NO₃), we traced N transformation processes, determined transformation rates, and assessed the roles of autotrophic nitrification, heterotrophic nitrification, denitrification, and co-denitrification in N₂O emissions. The specific objectives were to: (1) accurately measure gross N transformation rates in tobacco-cultivated soils; and (2) quantify the contributions of different pathways to N₂O production. We hypothesize that: (1) different organic substitution treatments lead to varied gross N transformations, resulting in distinct NH₄⁺ and NO₃⁻ production and consumption rates; and (2) different organic substitution treatments alter the relative importance of autotrophic nitrification, heterotrophic nitrification, denitrification, and co-denitrification as N₂O production pathways, ultimately leading to treatment-dependent variation in N₂O emissions.

Materials and methods

Information on soil samples

The field study was carried out in Qujing City (25°40' N, 103°39' E), Yunnan Province. The site has a mean annual temperature of 14.5 °C and a mean annual precipitation of 1,218.5 mm. According to the International Soil Texture Classification System, the tobacco-planting soil is categorized as a silty clay loam, comprising 27.2% clay, 6.5% sand, and 66.3% silt. For detailed methods used to determine soil physicochemical characteristics, refer to the [Supplementary Text S1](#).

The experiment followed a randomized block design comprising four nutrient management treatments, in which 15% of the mineral N input was substituted with organic manure, consistent with local conventional fertilization practices. Specifically, the four fertilization treatments were: mineral N fertilizer (SN); partial substitution of mineral N fertilizer with commercial organic manure (NC); partial substitution of mineral N fertilizer with farmyard manure (NF); and partial substitution of mineral N fertilizer with bio-organic manure inoculated with *Trichoderma viride* Pers. (NT). On May 2, 2023, tobacco plants of the 'Yunyan 100' variety were transplanted according to the local agricultural schedule, and harvesting was completed on September 10. The detailed fertilization program and

field management practices are shown in [Supplementary Tables S1 and S2](#). After harvest, soil was collected from the 0–20 cm cultivated layer and passed through a 2 mm stainless-mesh following manual removal of visible plant residues, roots, and gravel. Each soil sample was then split into two subsamples: one fraction was air-dried and stored at ambient laboratory temperature for physicochemical analysis, whereas the other portion was immediately frozen at –80 °C for subsequent ¹⁵N tracer experiment.

Laboratory ¹⁵N tracing incubation experiment

In this experiment, two sets of ¹⁵N-labeled treatments were applied to each soil sample: (1) ¹⁵NH₄⁺ treatment by adding ¹⁵NH₄NO₃ solution (10.23 ¹⁵N atom%), and (2) ¹⁵NO₃⁻ treatment by adding NH₄⁺¹⁵NO₃ solution (20.25 ¹⁵N atom%). Two criteria were applied to set the tracer abundance: (1) the atom% ¹⁵N in the targeted soil nitrogen pool must be approximately 5%–10% at 0.5 h after labeling; and (2) by the end of the incubation, the atom% ¹⁵N in all labeled pools remained measurably above the natural isotopic background. We selected the tracer addition volume to comply with standard soil incubation protocols and the specifications of the ¹⁵N isotope dilution method. The detailed procedures followed those described by Duan et al.^[24] and He et al.^[25]. By monitoring ¹⁵N dynamics across distinct soil N pools, we isolated N dynamics from confounding field variability and directly assessed how different treatments influence the soil's capacity to incorporate newly applied inorganic N.

Fresh soil, equivalent to 20 g on an oven-dry basis, was weighed and transferred into 120 mL serum vials, then pre-incubated in the dark at 25 °C for 48 h. After this acclimation period, 1 mL of either ¹⁵NH₄NO₃ or NH₄⁺¹⁵NO₃ solution was added with a pipette, supplying 50 mg N kg⁻¹ dry soil. Three independent replicates were prepared for each labeling treatment. Soil water content was regulated to 60% of field water-filled pore space, and vials were covered with perforated sealing, which limited evaporation while maintaining aerobic conditions. Incubations proceeded in the darkness at 25 °C, and destructive sampling was conducted at 0.5, 48, 96, and 144 h after tracer application. At each sampling time, soil NH₄⁺ and NO₃⁻ were extracted with 2 M KCl to determine their concentrations and ¹⁵N abundance. Following extraction of inorganic N, the residual was rinsed with deionized water to eliminate remaining mineral N, subsequently oven-dried, and then utilized for determination of organic N concentration and ¹⁵N abundance. Gaseous samples were collected at 48, 96, and 144 h. Before collection, the vials were tightly sealed and purged with mineral air (22% oxygen and 78% nitrogen) for 6 h to standardize the gas composition. Gas samples were then taken to assess N₂O concentration and ¹⁵N abundance. N₂O concentration was analyzed using a gas chromatograph (Agilent 7890A, Agilent Ltd, China), and the ¹⁵N abundance of NH₄⁺, NO₃⁻, organic N, and N₂O was analyzed with an isotope ratio mass spectrometer (Europa Scientific Integra, Crewe, UK)^[25]. For details on calculating N transformation rates and cumulative N₂O emissions contributions, see the [Supplementary Text S1](#).

N cycling functional gene abundances

DNA was extracted from soil samples at the end of incubation using the Fast DNA SPIN Kit for Soil (MP Biomedicals, USA), and its concentration and purity were assessed with a NanoDrop ND-1000 spectrophotometer (NanoDrop Technologies Inc., Wilmington, DE, USA), after which the extracts were diluted to 10 ng μL⁻¹ and their concentrations further verified on a StepOnePlus Real-Time PCR System (Applied Biosystems, USA). The assays targeted genes associated with N₂O emissions, including the archaeal and bacterial ammonia monooxygenase genes (AOA-*amoA* and AOB-*amoA*) associated with

nitrification, and the denitrification-related genes *nirS*, *nirK*, fungal-*nirK*, *nosZI*, and *nosZII*. Each qPCR mixture had a final volume of 20 μL , consisting of 10 μL of SYBR Green master mix (TaKaRa Biotech, Dalian, China), 0.4 μL of ROX reference dye, 2 μL of template DNA, 6.8 μL of nuclease-free water, and 0.4 μL of each primer (forward and reverse)^[24]. All amplifications were set up in triplicate for each sample, and a no-template control (nuclease-free water in place of DNA) was included. A standard curve was established using a plasmid harboring the target gene fragment, and a 10-fold serial dilution series (six points) was prepared for quantification. A single peak in the melting curve confirmed proper amplification. Copy numbers were log-transformed to normalize the data prior to statistical analysis. qPCR primers are detailed in [Supplementary Table S3](#).

Data analysis

Statistical analyses were conducted using one-way analysis of variance (ANOVA), and treatment means were subsequently separated with Duncan's multiple range test at a significance threshold of $p < 0.05$ in IBM SPSS Statistics 27 (IBM, Chicago, Illinois, USA). Pairwise linear relationships among variables were examined using Pearson's correlation analysis. Graphs were created using Origin v2021 and ChiPlot (www.chiplot.online). The model's optimal solution was found using the *fmincon* function in MATLAB v2021.

Results

Soil physicochemical characteristics

The one-way ANOVA results indicated that organic substitution treatments exerted significant effects on most soil physicochemical

characteristics, except for pH and DON ([Table 1](#)). No measurable parameter differed significantly ($p > 0.05$) between the NC and NT treatments. Compared with SN, the NT treatment resulted in markedly elevated SOC and TN contents, while the C/N ratio remained statistically unchanged. In contrast, the NF treatment exhibited the lowest TN content and the highest C/N ratio, both of which differed significantly from those of NT and NC. NF also had the highest DOC content, significantly exceeding that of the SN treatment ($p < 0.05$). Overall, organic substitution treatments improved soil physicochemical characteristics, with each treatment displaying specific strengths in particular indicators.

N transformation rates

[Table 2](#) presents the gross production and consumption rates of NH_4^+ and NO_3^- , along with their percentage contributions to the N transformation rates under different organic substitution treatments. Overall, gross NO_3^- production exceeded gross NH_4^+ production across all four fertilization treatments, while gross NH_4^+ consumption exceeded NO_3^- consumption. The gross NO_3^- production and NH_4^+ consumption rates ranked in the order: NF > NT > NC > SN, with NC, NF, and NT showing significantly higher values than SN, and significant differences were also observed among NC, NF, and NT ($p < 0.05$). Gross NO_3^- consumption rate followed the pattern: NT > NF > NC > SN, while gross NH_4^+ production rate followed the order: NT > NC > NF > SN. The NT treatment showed significantly elevated NO_3^- consumption and NH_4^+ production rates relative to the other treatments ($p < 0.05$).

In NO_3^- production, autotrophic nitrification was the primary contributing process, accounting for 65.0%, 63.4%, 63.0%, and 74.8% of the gross NO_3^- production across all four fertilization

Table 1 Soil physicochemical characteristics (mean \pm SD) under different organic substitution treatments

Treatment	pH	SOC (g kg ⁻¹)	TN (g kg ⁻¹)	C/N ratio	DOC (mg kg ⁻¹)	DON (mg kg ⁻¹)
SN	5.87 \pm 0.02 ^a	20.05 \pm 2.25 ^b	1.44 \pm 0.09 ^b	15.42 \pm 2.06 ^{ab}	21.29 \pm 1.06 ^b	24.37 \pm 1.67 ^a
NC	6.41 \pm 0.50 ^a	22.83 \pm 0.99 ^{ab}	1.80 \pm 0.16 ^a	12.78 \pm 1.44 ^b	25.68 \pm 0.13 ^{ab}	28.93 \pm 4.48 ^a
NF	6.45 \pm 0.49 ^a	23.57 \pm 2.08 ^{ab}	1.42 \pm 0.05 ^b	16.56 \pm 1.33 ^a	28.73 \pm 1.06 ^a	27.62 \pm 0.58 ^a
NT	6.33 \pm 0.52 ^a	25.79 \pm 0.20 ^a	1.99 \pm 0.14 ^a	13.03 \pm 0.84 ^b	25.54 \pm 4.26 ^{ab}	25.98 \pm 0.58 ^a

pH: soil potential of hydrogen; SOC: soil organic carbon; TN: total nitrogen; C/N ratio: SOC/TN ratio; DOC: dissolved organic carbon; DON: dissolved organic nitrogen; SN: synthesized chemical fertilizer; NC: commercial organic manure substituting 15% synthesized N; NF: farmyard manure substituting 15% synthesized N; NT: bio-organic (*Trichoderma viride* Pers.) manure substituting 15% synthesized N. Significant differences among treatments are denoted by different letters (one-way ANOVA, Duncan's multiple range test, $p < 0.05$).

Table 2 The average gross NH_4^+ and NO_3^- production and consumption rates, along with their percentage contributions to the total N pool production or consumption, under different organic substitution treatments at incubation times ranging from 0.5 to 144 h

N transformation rate	SN (mg N kg ⁻¹ soil d ⁻¹)	Contribution	NC (mg N kg ⁻¹ soil d ⁻¹)	Contribution	NF (mg N kg ⁻¹ soil d ⁻¹)	Contribution	NT (mg N kg ⁻¹ soil d ⁻¹)	Contribution
Gross NO_3^- production	6.00 \pm 0.04 ^d		6.43 \pm 0.14 ^c		7.43 \pm 0.13 ^a		7.21 \pm 0.04 ^b	
Autotrophic nitrification	3.90 \pm 0.13 ^c	65.0%	4.07 \pm 0.02 ^c	63.4%	4.68 \pm 0.14 ^b	63.0%	5.39 \pm 0.01 ^a	74.8%
Heterotrophic nitrification	2.10 \pm 0.17 ^c	35.0%	2.35 \pm 0.15 ^b	36.6%	2.74 \pm 0.07 ^a	37.0%	1.82 \pm 0.03 ^d	25.2%
Gross NO_3^- consumption	2.88 \pm 0.05 ^b		3.14 \pm 0.10 ^b		3.18 \pm 0.38 ^b		3.58 \pm 0.01 ^a	
DNRA	0.32 \pm 0.08 ^c	11.1%	0.47 \pm 0.05 ^b	15.1%	0.52 \pm 0.03 ^b	16.5%	1.25 \pm 0.02 ^a	34.9%
NO_3^- immobilization	2.54 \pm 0.04 ^a	88.1%	1.82 \pm 0.10 ^c	57.8%	2.19 \pm 0.05 ^b	68.7%	2.27 \pm 0.05 ^b	63.3%
Gross NH_4^+ production	0.88 \pm 0.04 ^d		1.21 \pm 0.04 ^b		1.01 \pm 0.03 ^c		1.37 \pm 0.01 ^a	
DNRA	0.32 \pm 0.08 ^c	36.4%	0.47 \pm 0.05 ^b	39.3%	0.52 \pm 0.03 ^b	51.8%	1.25 \pm 0.02 ^a	90.9%
Mineralization	0.56 \pm 0.04 ^b	63.6%	0.73 \pm 0.03 ^a	60.7%	0.49 \pm 0.06 ^c	48.2%	0.13 \pm 0.00 ^d	9.1%
Gross NH_4^+ consumption	6.14 \pm 0.10 ^d		6.50 \pm 0.03 ^c		6.96 \pm 0.02 ^a		6.81 \pm 0.02 ^b	
Autotrophic nitrification	3.89 \pm 0.14 ^c	63.5%	4.07 \pm 0.02 ^c	62.6%	4.68 \pm 0.14 ^b	67.3%	5.39 \pm 0.01 ^a	79.2%
NH_4^+ immobilization	1.30 \pm 0.03 ^a	21.0%	1.16 \pm 0.12 ^{ab}	17.9%	1.07 \pm 0.05 ^b	15.4%	1.17 \pm 0.04 ^{ab}	17.2%

Nitrate nitrogen: NO_3^- ; ammonium nitrogen: NH_4^+ ; dissimilatory nitrate reduction to ammonium: DNRA; SN: synthesized chemical fertilizer; NC: commercial organic manure substituting 15% synthesized N; NF: farmyard manure substituting 15% synthesized N; NT: bio-organic (*Trichoderma viride* Pers.) manure substituting 15% synthesized N. Significant differences among treatments are denoted by different letters (one-way ANOVA, Duncan's multiple range test, $p < 0.05$).

treatments. It was also the main process for NH_4^+ consumption, contributing 63.5%, 62.6%, 67.3%, and 79.2% of the gross NH_4^+ consumption across all four fertilization treatments. Autotrophic nitrification in the NF and NT treatments was significantly elevated relative to that observed in the SN treatment.

DNRA and NO_3^- immobilization contributed over 70% to NO_3^- consumption, with NO_3^- immobilization as the main process. In contrast, DNRA was the primary source of NH_4^+ production in the NF and NT treatments, contributing 51.8% and 90.9%, respectively, while mineralization was the dominant process in the SN and NC treatments, contributing 63.6% and 60.7%, respectively.

N₂O emission pathways

Figure 1a shows that the N_2O emission rates remained dynamic at 60% of the field water-filled pore space, with a significant emission peak on the fourth day of the labeling experiment. Among the four fertilization treatments, the cumulative N_2O emissions were, in descending order, $\text{SN} > \text{NF} > \text{NC} > \text{NT}$, with organic substitution treatments producing significantly lower emissions than the SN treatment ($p < 0.05$), whereas differences among the organic substitution treatments were not statistically discernible (Fig. 1b). Relative to the SN treatment, the organic substitution treatments also attenuated the cumulative N_2O emissions from autotrophic nitrification. However, these treatments had varying effects on N_2O emissions from denitrification and heterotrophic nitrification. NF and NC increased N_2O emissions during denitrification, while NT decreased them. NC increased N_2O emissions during heterotrophic nitrification, whereas NF and NT reduced them.

Throughout the incubation period, the relative contribution of autotrophic nitrification to N_2O emission pathways increased over incubation time, whereas the proportional roles of heterotrophic nitrification and denitrification decreased (Fig. 2). On average, autotrophic nitrification accounted for 67.2%, 62.3%, 65.4%, and 59.7% of N_2O emissions under the four fertilization treatments; heterotrophic nitrification contributed 15.3%, 16.6%, 15.8%, and 18.6%, respectively; and denitrification contributed 17.5%, 20.1%, 18.8%, and 21.7%, respectively. Co-denitrification made no significant contribution to N_2O emissions.

Abundance of N₂O-related functional genes

Organic substitution treatments significantly affected the copy numbers of nitrification and denitrification functional genes in tobacco soil

(Fig. 3). Relative to the SN treatment, the NF treatment enhanced the abundance of the AOA-*amoA* gene by 2.7%, and *nosZII* gene abundance by 7.4% ($p < 0.05$). However, under the NC and NT treatments, *nirK* gene abundance decreased by 2.4% and 3.1%, respectively, relative to SN ($p < 0.05$). Under the three organic substitution treatments, NF consistently supported higher abundances of AOA-*amoA*, *nirK*, and *nosZII* than either NC or NT, and AOB-*amoA* gene abundance was also greater in NF than NT ($p < 0.05$). Additionally, *nosZII* gene abundance differed significantly among all four fertilization treatments ($p < 0.05$). No significant differences were detected for *nirS*, fungal-*nirK*, and *nosZI* gene abundances across the treatments.

Inter-group correlation analysis

Soil biochemical characteristics influenced N transformation rates and N_2O emission pathways in different patterns. During N transformations (Fig. 4a), DOC showed a significant negative relationship with both gross NO_3^- production and NH_4^+ consumption, whereas SOC and fungal-*nirK* were negatively associated with gross NO_3^- consumption and NH_4^+ production ($p < 0.05$). In contrast, the *nirK* gene exhibited a positive correlation with gross NO_3^- consumption and NH_4^+ production. In addition, the C/N ratio and *nirS* were positively related to gross NH_4^+ production ($p < 0.01$). Regarding N_2O emission pathways (Fig. 4b), SOC, TN, and fungal-*nirK* were positively correlated with cumulative N_2O emissions from autotrophic nitrification ($p < 0.01$), whereas the C/N ratio, *nirS*, and *nirK* genes were negatively correlated ($p < 0.05$).

Discussion

Regulatory effects on N transformation processes

Our study used a ^{15}N tracer addition rate of 50 mg N kg^{-1} soil, which exceeds typical *in situ* inorganic N pools. This rate was chosen to ensure analytical precision for estimating gross N transformation rates, consistent with previous ^{15}N tracer studies^[24]. We acknowledge that this relatively high N input may stimulate microbial N processes and thereby lead to overestimation of absolute gross rates relative to field conditions; therefore, our results should be interpreted as potential, rather than strictly *in situ* rates. Nevertheless, because all treatments

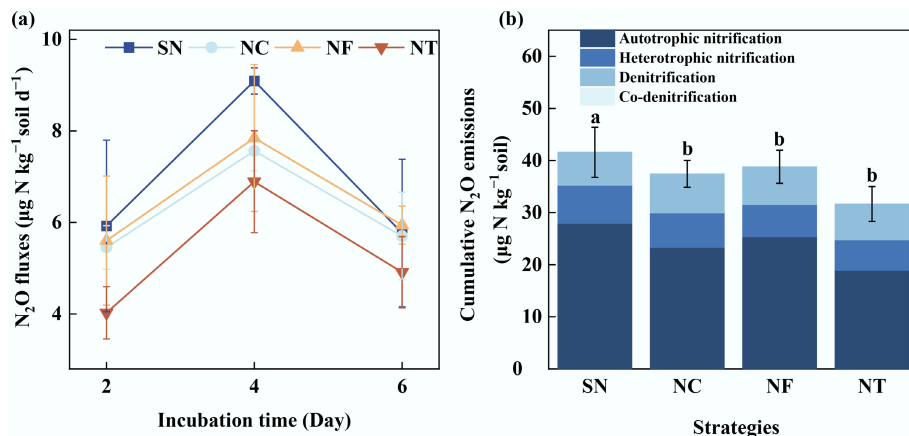


Fig. 1 (a) N_2O fluxes, and (b) cumulative N_2O emissions via four pathways in tobacco fields under organic substitution treatments. SN: synthesized chemical fertilizer; NC: commercial organic manure substituting 15% synthesized N; NF: farmyard manure substituting 15% synthesized N; NT: bio-organic (*Trichoderma viride* Pers.) manure substituting 15% synthesized N. Significant differences among treatments are denoted by different letters (one-way ANOVA, Duncan's multiple range test, $p < 0.05$).

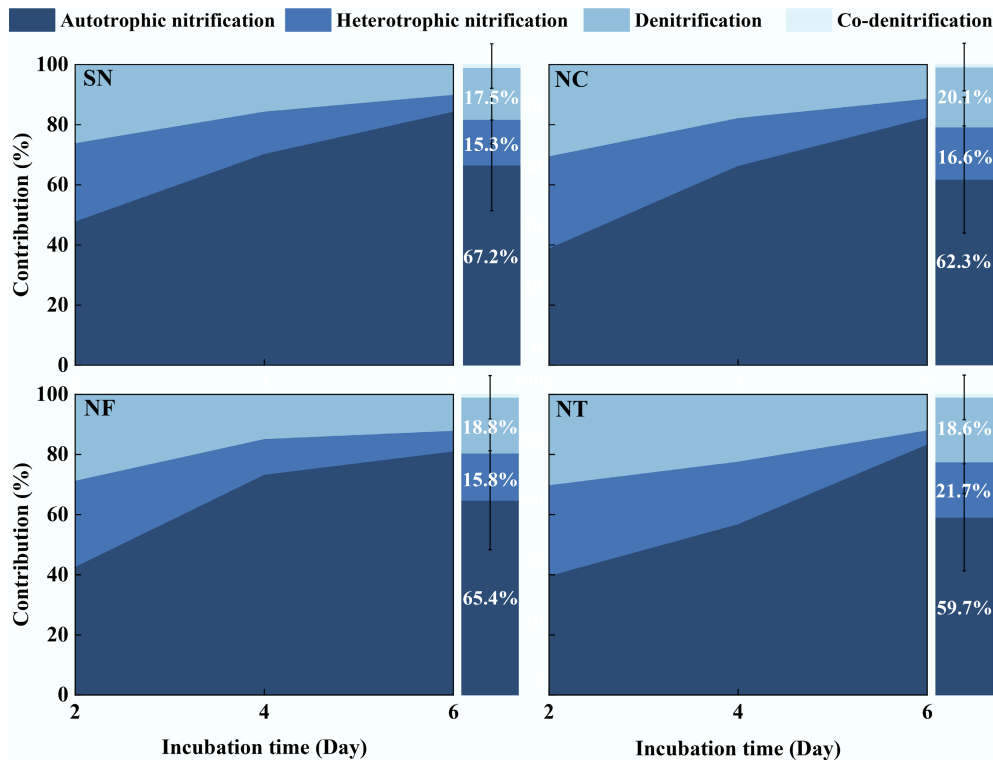


Fig. 2 N₂O emission pathways and their average contributions in tobacco fields under organic substitution treatments. SN: synthesized chemical fertilizer; NC: commercial organic manure substituting 15% synthesized N; NF: farmyard manure substituting 15% synthesized N; NT: bio-organic (*Trichoderma viride* Pers.) manure substituting 15% synthesized N.

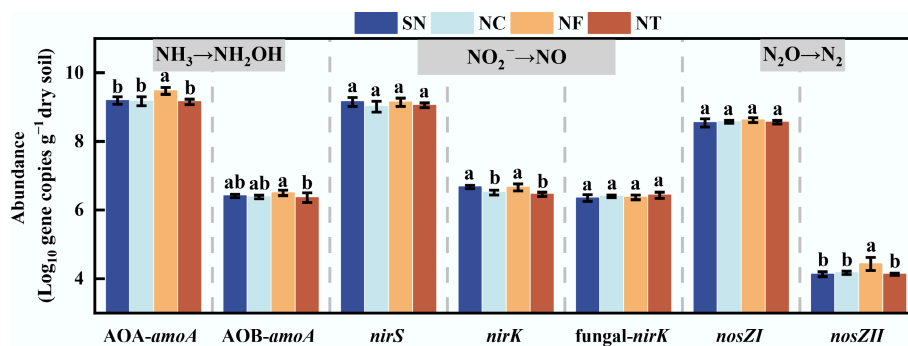


Fig. 3 Log₁₀-transformed gene copy numbers of soil functional genes (AOA-amoA, AOB-amoA, nirS, nirK, fungal-nirK, nosZI, and nosZII) after 144 h of incubation under different organic substitution treatments. SN: synthesized chemical fertilizer; NC: commercial organic manure substituting 15% synthesized N; NF: farmyard manure substituting 15% synthesized N; NT: bio-organic (*Trichoderma viride* Pers.) manure substituting 15% synthesized N. Significant differences among treatments are denoted by different letters (one-way ANOVA, Duncan's multiple range test, $p < 0.05$).

received the same tracer addition under identical conditions, any systematic bias should be consistent across treatments, and the observed treatment differences and main conclusions remain valid.

In tobacco fields in Yunnan, the gross rate of NO₃⁻ production exceeded that of NH₄⁺ production, while the consumption of NH₄⁺ was generally higher than the consumption of NO₃⁻ (Table 2). These patterns indicate an intensive nitrification process and a rapid turnover of the NH₄⁺ pool, rather than a simple accumulation of mineral N. This pattern is closely linked to the ecological conditions of tobacco-growing regions. Tobacco is a typical nitrate-preferring crop, and its fertilization and quality control rely primarily on NO₃⁻ as the N source^[26]. Once NH₄⁺ is produced in the soil, this NH₄⁺ is rapidly nitrified under the warm, well-aerated conditions of Yunnan, continuously replenishing the NO₃⁻ pool. Meanwhile, tobacco has a

low tolerance for NH₄⁺, and therefore maintaining high NH₄⁺ levels in the rhizosphere is not desirable^[27]. As a result, mineralization provides only limited NH₄⁺, and NH₄⁺ is quickly consumed through nitrification and microbial assimilation, so that its consumption exceeds its production. In contrast, NO₃⁻ is continuously replenished by nitrification and fertilization, and its supply often exceeds its removal via denitrification and leaching, leading to a net surplus of NO₃⁻.

This investigation revealed that organic substitution treatments significantly affected N production and consumption processes, as evidenced by quantified N transformation rates. Relative to the SN treatment, treatments involving organic substitution produced substantially greater gross NO₃⁻ and NH₄⁺ production and consumption. This indicates that the organic matter and associated

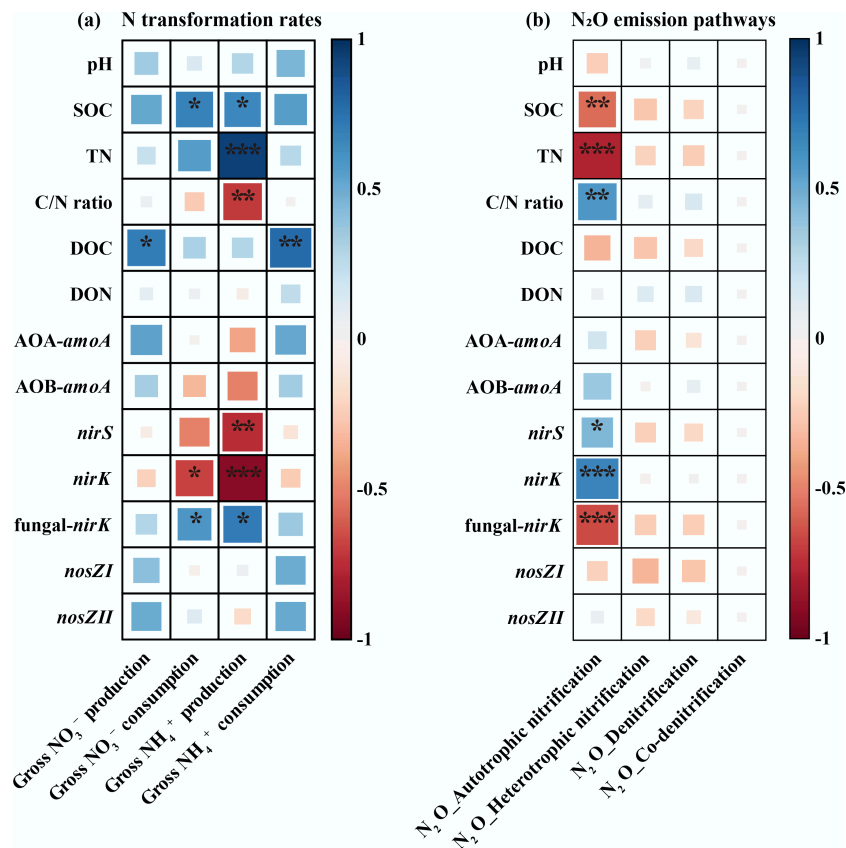


Fig. 4 Heatmap of Pearson's correlation matrix between soil biochemical characteristics and gross NH_4^+ and NO_3^- (a) production and consumption rates, and (b) cumulative N_2O emissions via pathways in tobacco soils. SN: synthesized chemical fertilizer; NC: commercial organic manure substituting 15% synthesized N; NF: farmyard manure substituting 15% synthesized N; NT: bio-organic (*Trichoderma viride* Pers.) manure substituting 15% synthesized N. * indicates significance at $p < 0.05$; ** indicates significance at $p < 0.01$; *** indicates significance at $p < 0.001$.

microorganisms delivered by organic manures enhance soil N supply capacity by stimulating microbial biomass and metabolic activity, thereby accelerating the transformation of inorganic N pools^[8,28]. The NT treatment showed the most significant improvement in gross NH_4^+ production and NO_3^- consumption rates, indicating that bio-organic manure inputs provide readily available C and N and enrich beneficial microbial communities, leading to faster production and consumption of NO_3^- and NH_4^+ . Additionally, improvements in soil aggregation, porosity, moisture retention, and pH buffering create more favorable and stable micro-environments for microbial functioning, sustaining rapid transformation of these pools^[29]. In contrast, the sole mineral inputs under SN treatment likely constrained microbial biomass through C limitation and soil acidification, dampening community activity and slowing N transformations.

Further examination of N transformation processes showed that autotrophic nitrification contributed more to NO_3^- production and NH_4^+ consumption, particularly under the NF and NT treatments, where nitrification rates were markedly elevated relative to those in the NC and SN treatments (Table 2). However, the copy numbers of AOA-*amoA* and AOB-*amoA* genes under NT treatment were statistically indistinguishable from those in the SN treatment (Fig. 3), suggesting that the enhancement of autotrophic nitrification results from higher ammonia-oxidizing rates per cell or per *amoA* gene copy, rather than from increased *amoA* gene abundance. Functional microbes and their metabolites in bio-organic manure help optimize soil microaerophilic niches and suppress pathogenic and

heterotrophic microorganisms, thereby reducing competition for oxygen and NH_4^+ . Collectively, these processes substantially strengthen the ammonia-oxidizing capacity of both AOA and AOB, thereby amplifying their overall role in NO_3^- production^[30]. In the present study, despite the higher copy numbers of AOA-*amoA* and AOB-*amoA* genes observed under the NF treatment, their nitrification rates were lower than those measured under NT. This apparent discrepancy is probably attributable to the high organic matter content and incompletely decomposed residues in farmyard manure, which favored heterotrophic microbes and immobilization processes. As autotrophic nitrifiers preferentially utilize inorganic NH_4^+ rather than organic N, a slower mineralization and a stronger microbial competition for inorganic N could restrict their effective substrate supply^[31]. Therefore, even with higher *amoA* gene abundance, ammonia oxidation rates in NF were not significantly enhanced.

The application of organic manures enriched the content of labile organic matter in the soil, thereby supplying ample C sources for microbial metabolism and promoting ammonia oxidation as well as the DNRA process^[32]. Heterotrophic microorganisms contribute to NO_3^- production during organic matter decomposition, activating the native microbial community and increasing heterotrophic nitrifier diversity, which boosts heterotrophic nitrification rate^[33]. Consequently, the combined enhancement of autotrophic and heterotrophic nitrification in treatments involving organic substitution markedly accelerated the transformation of NH_4^+ to NO_3^- . It is noteworthy that the NF and NT treatments altered the dominant

source of NH_4^+ from organic N mineralization to DNRA. Under the NT treatment, DNRA contributed up to 90.9% of NH_4^+ production, far exceeding the levels reported for most aerobic soils (Table 2). However, this very high contribution ratio mainly reflects the relatively low gross NH_4^+ production and weak mineralization rate under NT, rather than an abnormally high DNRA rate; DNRA operated at a normal rate but accounted for most of the limited NH_4^+ generated. Mechanistically, the bio-organic manure likely stimulated nitrification, so that newly produced NH_4^+ was rapidly oxidized to NO_3^- and NH_4^+ accumulation was suppressed^[34]. It also increased the bulk soil C/N ratio and DOC, which stoichiometrically favors NO_3^- reduction via DNRA over organic N mineralization as a source of NH_4^+ ^[35]. Although the contribution rate of DNRA increased substantially, the DNRA transformation rate measured in this study (0.32–1.25 mg N kg⁻¹ soil d⁻¹) remained within the typical global range (0.18–1.30 mg N kg⁻¹ soil d⁻¹) and was consistent with existing data for Fujian red soil (0.70–2.38 mg N kg⁻¹ soil d⁻¹)^[36,37], indicating that the DNRA transformation rate under our experimental conditions is reasonable at both regional, and global scales.

Variations in N₂O emission pathways and magnitudes

N₂O is a highly effective radiatively active gas, and its generation is primarily controlled by nitrification and denitrification processes^[38]. In the present incubation, isotopic partitioning indicated that autotrophic nitrification is the main N₂O emission pathway, accounting for over 60% of total emissions (Fig. 2). Compared to SN, all organic substitution treatments exhibited lower cumulative N₂O emissions in this assay (Fig. 1). This result is consistent with previous field studies, which have concluded that organic substitution can significantly reduce N₂O emissions^[39]. Specifically, these treatments effectively reduce N₂O emissions during autotrophic nitrification. A plausible explanation is that organic manures typically supply N in more slowly mineralizable forms, which reduce short-term N accumulation and prevent excessive soil N concentrations, inhibiting nitrification by autotrophic bacteria and slowing N₂O production and emissions^[40]. Under plant-free conditions, although the NT treatment resulted in a higher autotrophic nitrification rate, the application of bio-organic manure improved soil structure and aeration. As a result, nitrification occurs mainly under well-oxygenated, stable conditions, which reduces the fraction of N₂O produced during nitrification^[41]. At the same time, bio-organic manure increases SOC content and creates localized anaerobic micro-environments within soil aggregates. These localized anaerobic niches intensify denitrification activity and facilitate the further reduction of N₂O to N₂, thereby further consuming N₂O that has already been formed. Accordingly, the contributions of heterotrophic nitrification and denitrification to N₂O emissions were significantly reduced under NT, and the abundance of *nirK* decreased (Fig. 3). Because *nirK* encodes nitrite reductase in the denitrification pathway, its lower abundance indicates a reduced potential for denitrification-derived N₂O^[42]. Additionally, the microbial agents and appropriate C/N ratio in bio-organic manure may have promoted the dominance of competitive micro-organisms during autotrophic nitrification, further reducing N₂O emissions from heterotrophic nitrification and denitrification by inhibiting microbial activity during anaerobic N transformation^[21]. The NF treatment reduced N₂O emissions during heterotrophic nitrification and significantly increased *nosZII* gene abundance, promoting N₂O reduction to N₂ and effectively lowering N₂O emissions^[43]. In contrast, the NC treatment led to higher N₂O emissions during heterotrophic nitrification and denitrification and significantly reduced *nirK* gene abundance, suggesting suppressed denitrifier activity. Under the NC

treatment, soil exhibited a low C/N ratio and elevated DON (Table 1). Commercial organic manure, although formulated and relatively stable in terms of its bulk organic fraction, often contains a substantial proportion of readily available N and labile C, which can be quickly mineralized and denitrified^[44]. This localized pulse of high DON combined with a relatively low C/N ratio in the decomposing manure can promote rapid N release and the formation of transient anaerobic microsites, thereby increasing N₂O production, even though *nirK* gene abundance is relatively low. Thus, N₂O emission pathways and intensity are closely linked to the fertilization treatment. Over-reliance on chemical N fertilizers increases N₂O emissions, while organic substitution treatments can effectively regulate N₂O emission pathways, achieving more eco-friendly fertilization goals.

Relationships among soil biochemical characteristics, N transformation rates, and N₂O emission pathways

This study investigated the interaction of soil biochemical characteristics in N transformations in tobacco soils. Organic substitution treatments significantly improved soil physicochemical characteristics, particularly SOC and TN contents (Table 1). Compared with the SN treatment, NT increased the contents of SOC and TN, indicating that bio-organic manure, by supplying ample C and N, promoted organic matter accumulation and enhanced soil fertility^[45]. This likely boosted microbial activity and N metabolism by facilitating N transformations^[42]. Correlation analysis showed that SOC and TN contents were negatively associated with gross NH_4^+ production rate, but positively correlated with cumulative N₂O emissions from autotrophic nitrification (Fig. 4). High SOC and TN enhanced microbial activity, leading to competitive consumption of NH_4^+ and oxygen by micro-organisms, which decreased NH_4^+ production and increased N₂O emissions^[46]. Additionally, the C/N ratio was positively associated with gross NH_4^+ production rate, but negatively associated with cumulative N₂O emissions from autotrophic nitrification. A high C/N ratio indicates more SOC and less TN, promoting the mineralization of organic N into NH_4^+ . However, a high C/N ratio also implies N deficiency, weakening nitrification and reducing N₂O emissions^[47]. Although the NF treatment had the lowest TN content, its higher DOC and C/N ratio indicated that farmyard manure might contain more recalcitrant organic matter, reducing mineralization and potentially influencing N release and microbial preference for C sources^[48]. This elevated C/N ratio could suppress rapid ammonium release, thereby affecting nitrification rates and ultimately decreasing N losses^[49].

In addition, the abundances of *nirS* and *nirK* genes were associated with gross NH_4^+ production rate, but negatively related to cumulative N₂O emissions from autotrophic nitrification. High expression of these genes indicated efficient bacterial denitrification, promoting the N cycle, increasing NH_4^+ production, and reducing N₂O emissions^[50]. Conversely, the abundance of fungal-*nirK* genes showed a negative correlation with gross NH_4^+ production rate, and a positive correlation with cumulative N₂O emissions from autotrophic nitrification. High fungal-*nirK* gene abundance resulted in lower denitrification efficiency, reducing NH_4^+ production, and increasing N₂O emissions^[51].

In summary, organic substitution treatments improved soil nutrients and microecological stability by increasing both C and N reserves, adjusting the C/N balance, and enriching microbial C substrates. This strengthened N transformation processes and reduced N₂O emissions, offering insights into the complexity of soil N cycling and its associated gaseous losses.

Conclusions

This study systematically examined how organic substitution treatments influence gross N transformation processes and N₂O emission pathways in tobacco soils. The findings demonstrate that, relative to the SN treatment, organic substitution markedly enhanced NO₃⁻ formation and increased both the production and consumption rates of NH₄⁺. Autotrophic nitrification dominated NO₃⁻ production and NH₄⁺ consumption, particularly in the NF and NT treatments. Additionally, these treatments substantially reduced cumulative N₂O emissions, although autotrophic nitrification remained the primary N₂O emission source. In summary, organic substitution treatments modulate N transformation processes and significantly reduce N₂O emissions, supporting sustainable agriculture and greenhouse gas mitigation. These findings provide crucial theoretical backing and practical guidance for adopting organic substitution treatments to enhance soil quality and reduce agricultural greenhouse gas emissions.

Supplementary information

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Author contributions

The authors confirm their contributions to the paper as follows: Bingxue Wang: formal analysis, investigation, writing – original draft; Xiaopeng Deng: data curation, resources; Zhonglong Wang: investigation, data curation; Bin Wang: investigation, data visualization; Ruibao Wang: investigation, resources; Yali Zhang: investigation, writing – review & editing; Zhengqin Xiong: conceptualization, funding acquisition, supervision, writing – review & editing. All authors reviewed the results and approved the final version of the manuscript.

Data availability

The data generated and analyzed during the current study are not publicly archived but can be obtained by the corresponding author upon reasonable request.

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Declarations

Competing interests

The authors declare that they have no conflict of interest.

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